

### IN THE CLAIMS

Please amend the claims as follows:

1. (Currently amended) A method for the preparation of a cathode active material comprising:  
mixing, milling and sintering materials for synthesis of a compound represented by the general formula  $\text{Li}_x\text{FePO}_4$ , where  $0 < x \leq 1$ , and adding a carbon material at an optional time point in the course of said mixing, milling and sintering;  
employing  $\text{Li}_3\text{PO}_4$  and  $\text{Fe}_3(\text{PO}_4)_2$  or its hydrate  $\text{Fe}_3(\text{PO}_4)_2 \cdot n\text{H}_2\text{O}$ , where n denotes the number of hydrates, as the materials for synthesis of said  $\text{Li}_x\text{FePO}_4$ ; and  
setting the oxygen concentration in a sintering atmosphere to greater than ~~zero~~ or equal to 3ppm, but less than or equal to 1012 ppm in volume at the time point of sintering.
2. (Currently amended) A method for the preparation of a non-aqueous electrolyte cell including a cathode having a cathode active material, an anode having an anode active material and a non-aqueous electrolyte, wherein  
in preparing said cathode active material, sintering starting materials for synthesis of a compound represented by the general formula  $\text{Li}_x\text{FePO}_4$ , where  $0 < x \leq 1$ , are mixed, milled and a carbon material is added at an optional time point in the course of said mixing, milling and sintering;  
 $\text{Li}_3\text{PO}_4$  and  $\text{Fe}_3(\text{PO}_4)_2$  or its hydrate  $\text{Fe}_3(\text{PO}_4)_2 \cdot n\text{H}_2\text{O}$ , where n denotes the number of hydrates, are used as the starting materials for synthesis of said  $\text{Li}_x\text{FePO}_4$ ; and  
the oxygen concentration in a sintering atmosphere is set to greater than ~~zero~~ or equal to 3ppm, but less than or equal to 1012 ppm in volume at the time point of sintering.
3. (Previously amended) The method for the preparation of a non-aqueous electrolyte cell according to claim 2 wherein said non-aqueous electrolyte comprising a non-aqueous electrolyte includes a dissolved electrolyte in a non-aqueous solvent.

4. (Original) The method for the preparation of a non-aqueous electrolyte cell according to claim 2 wherein said non-aqueous electrolyte is a solid electrolyte.

5. (Original) The method for the preparation of a non-aqueous electrolyte cell according to claim 2 wherein said anode is a material capable of doping/undoping lithium.

6. (Original) The method for the preparation of a non-aqueous electrolyte cell according to claim 2 wherein said anode is a carbon material.